

CA

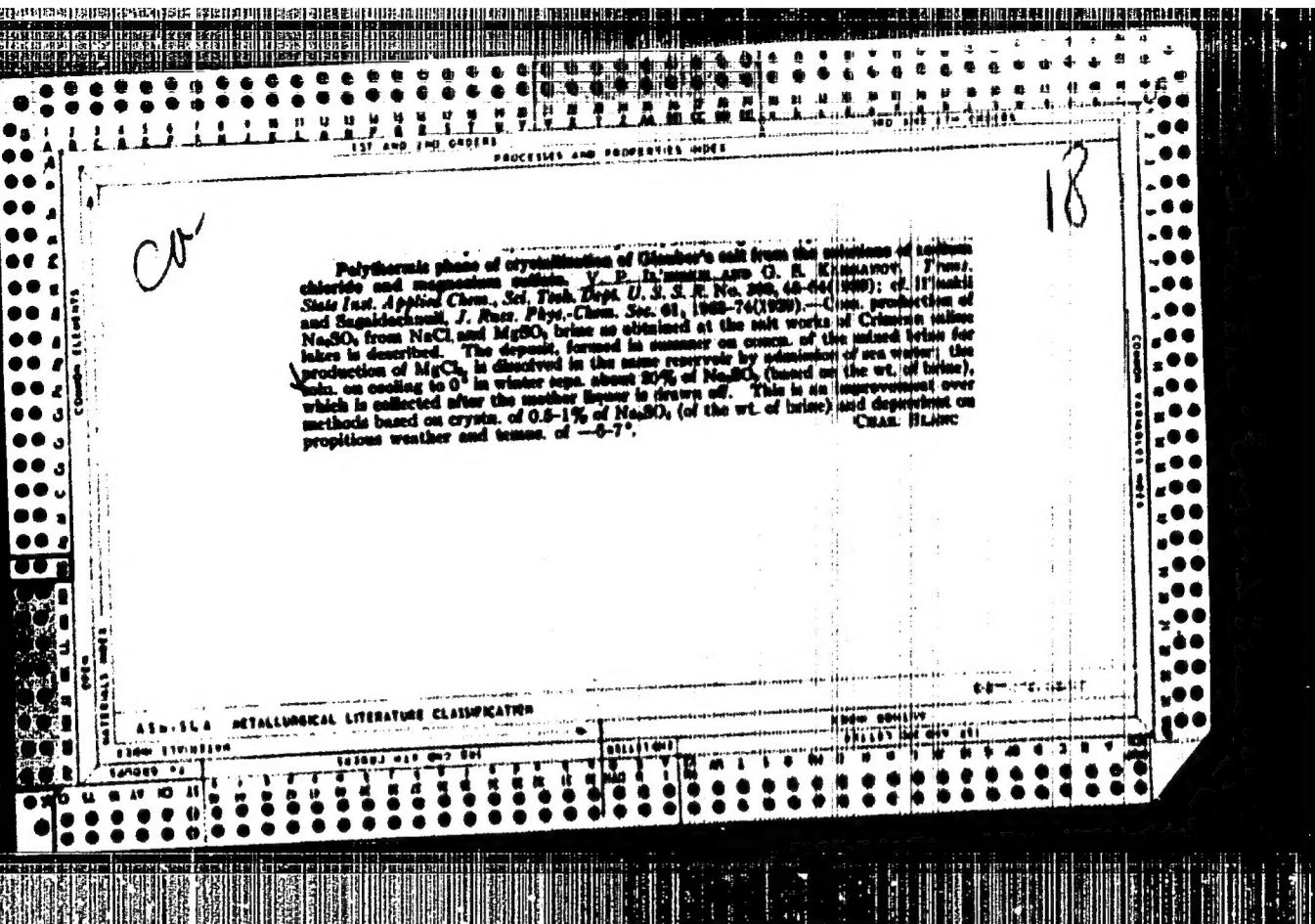
Z

The polymorphic field of crystallization of Glauber salt in the systems  $2\text{NaCl} + \text{MgSO}_4$  and  $\text{Na}_2\text{SO}_4 + \text{MgCl}_2$ . V. E. MIRONOV AND A. P. SHOKHINAKOV. *J. Russ. Phys.-Chem. Soc.* 64, 1953-74 (1932).—In the temp. limits  $+32.6^\circ$  to  $-12^\circ$  a considerable increase in the field of crystals of Glauber salt takes place with lowering of temp., mainly at the expense of the crypta of Glauber salt from  $+32.6^\circ$  to  $+20^\circ$  takes place the increase takes place at the expense of NaCl, astrakanite and  $\text{MgSO}_4 \cdot \text{H}_2\text{O}$ . Beginning with  $-1.2^\circ$  the crystal. field of Glauber salt decreases rapidly with fall of temp., since a new field of  $\text{H}_2\text{O}$  appears. The displacement of the triple point  $111^\circ$ , from  $+25^\circ$  to  $-8^\circ$ , runs parallel with the diagonal  $\text{Na}_2\text{SO}_4 - \text{MgCl}_2$ . S. L. MIRONOV

AIA-164 METALLURGICAL LITERATURE CLASSIFICATION

LOW TEMPERATURE

1930-52 MEL ONLY ONE



4  
C  
Electrodeposition of chromium. V. P. IL'INOV, N. P. LAPIN AND L. N. KOLTS  
Zhur. Priborostroeniye i Khim. 3, 308-310 (1980).<sup>21</sup> The increase in concn. of sulfate ions (above 3 g. of Cr<sub>2</sub>(SO<sub>4</sub>)<sub>2</sub> per L) decreases the current efficiency, improves theuster of the deposited metal, increases to a certain extent the permissible working interval between the min. and max. allowable c. d., and probably increases the hardness of the deposits. For protection of Fe from corrosion the layer of Cr must be above 0.01 mm. thick. The Cr films are obtained which are more resistant to corrosion than the Cr films deposited from saline, with smaller concn. of sulfate ions. The data are calculated so as to permit the selection of the operating conditions for the particular type of the Cr plate, which is desired for covering the metal surface. V. KALITINOVSKY

APPENDIX METALLURGICAL LITERATURE CLASSIFICATION

**Electrolytic production of potassium ferricyanide.** V. P. LINDSEY, Am. N. 11, LAPIN, Zhur. Prilobnost. Khim., 3, 691-695 (1900).—A mix. of  $K_4[Fe(CN)_6]$  is electrolyzed at 65° between the Fe cathode and Ni anode. The anode current is 5 amp., and at the cathode 20 to 30 amp., with, and 160-180 amp./sq. in. without, the anode dia-phragm. The dia-phragm, at the cathode, increases the current output, but it soon deteriorates because of the mech. action of H<sub>2</sub>. Addition is rendered difficult by treating it with HCl and boiling at 100°-110°. If the distance is rendered small enough, less than 4 cm., the distribution of current at the anode is irregular. Voltage is 2.0-3.2 v. with, and 3.2-3.9 v. without, dia-phragm.  $K_4[Fe(CN)_6]$  acts on the Ni anode and if present should be separated from  $K_4[Fe(CN)_6]$  by recryst. Electrolysis is discontinued when the  $K_4[Fe(CN)_6]$  conc. falls to 15-20 g./L.  $K_4[Fe(CN)_6]$  is crystal. From the anode, pooled to 15-20%. The mother liquor is again mixed with  $K_4[Fe(CN)_6]$  at 50° and the process repeated. After the 4th cycle the anode is evapd. to dryness. KOH thus obtained contains a certain quantity of CN complex, but it might be utilized for ads.  $K_4[Fe(CN)_6]$ . A flow sheet of the process based on mass-curr. scale ratio, is appended. The following poly. dia-phragms are given: mixts. of  $K_4[Fe(CN)_6]$  and  $K_4[Fe(CN)_6$  at 25°, 40°, 55° and at 10° mixts. of  $K_4[Fe(CN)_6$  and  $K_4[Fe(CN)_6]$  in water, 1 N and 2 N KOH at 25°, 40° and 10°. V. P. L.

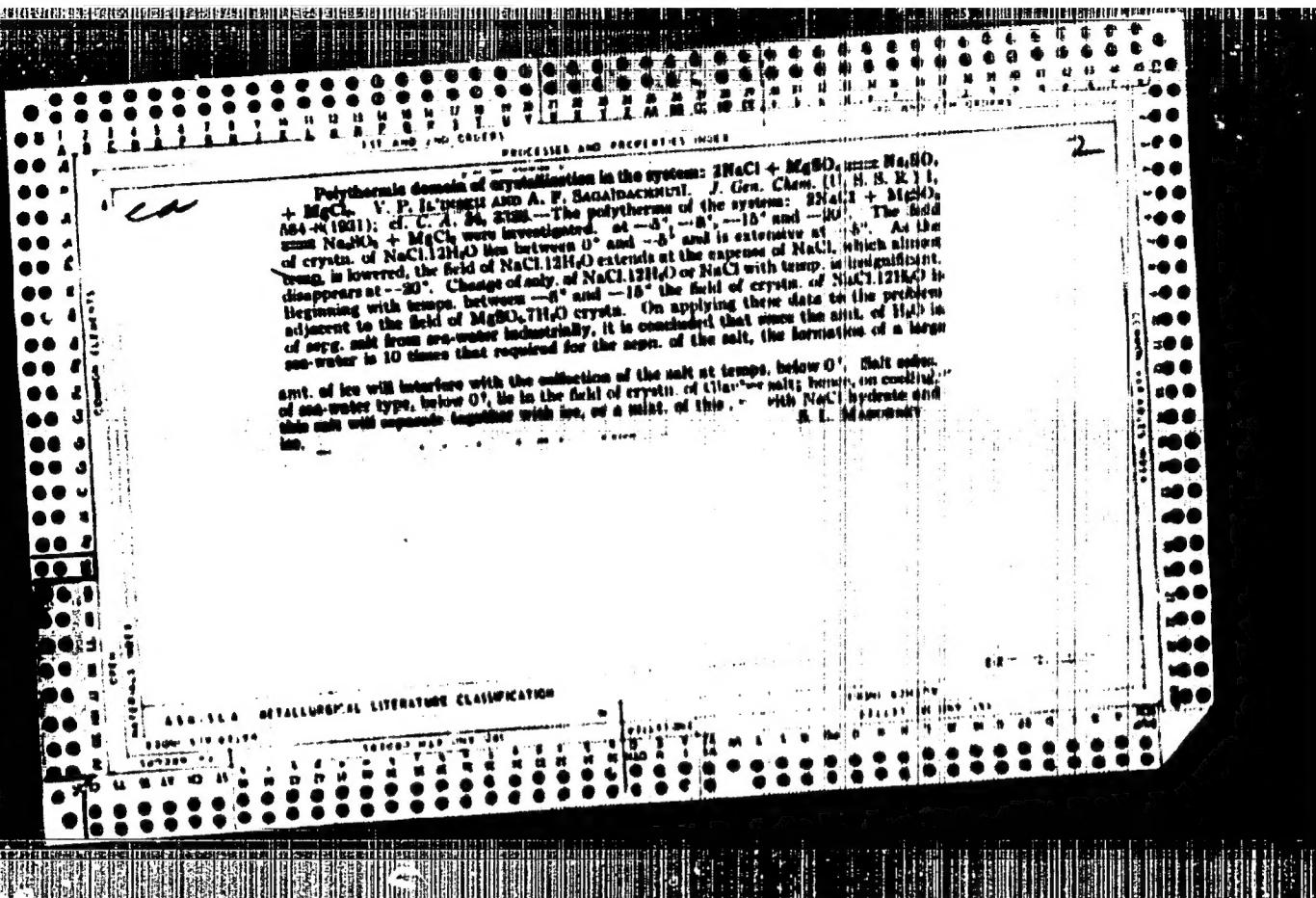
ALFA-SEA METALLURGICAL LITERATURE CLASSIFICATION

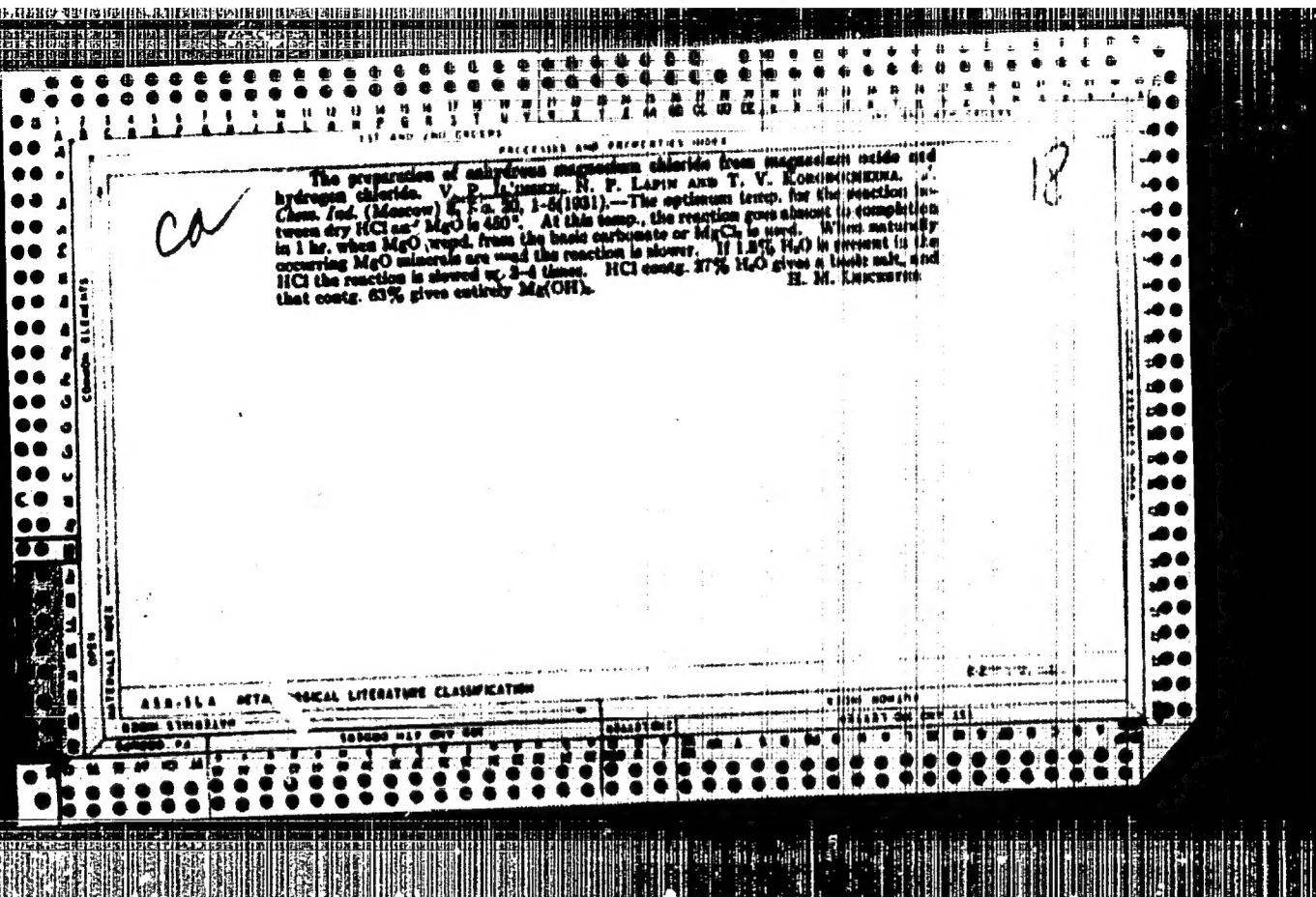
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Car 46

REVIEWED AND APPROVED BY  
THE DIRECTOR OF CENTRAL INTELLIGENCE WITH DRAFTING INSTRUCTIONS  
Electrolytic production of ammonium pernitrate with high current densities  
A. P. L'vovskii and O. V. Gulyayev. Zav. Pochtovo-Telegraf. Akad. Nauk SSSR. App.  
is described which employs a c.d. of 600-700 amp/m<sup>2</sup> in at the cathode. V. K.





Preparation of manganese dioxide by electrolysis. V. I. B. [initials] App. N. F. LAPIN. J. Applied Chem. (U. S. S. R.) 4, 757-64 (1951). Pyrolysis was mixed with anthracite and reduced at 800-900° to  $MnO_2$ . The latter, after the removal of impurities, was dissolved in dil.  $H_2SO_4$ . The electrolyte is carried out at about 25° and the e. d. on the anode should not exceed 8 amp. per sq. dm. and on the cathode 20 amp. per sq. dm. The cathode was made of C rods and the anode of lead. The  $H_2SO_4$  concn. was about 25% and the potential drop 2-3 d. v. The total yields of  $MnO_2$  were 75-85%. The product has the appearance of small shiny (luster) and is a very active depolarizer for galvanic cells. A number of experiments were carried out with  $H_2SO_4$  electrolyte and Zn electrodes, which were of little success, as well as the expts. which make it advisable to start the process on a semi-manufacturing scale are reported in detail. A. A. [initials]

PROCESSING AND PREPARATION  
Production of aluminum oxide from titanium boride  
by the combined pyrolytic methods. V. P. NIKONOV  
Ind. Applied Chem. (U. S. S. R.) 1953, No. 10, 40-70.  
The procedure is based on the 2 methods of pyrolysis  
(C. A. 46, 2626) and titanium pentoxide (1950), 40,000 and 347,262  
on the reaction:  $\text{Na}_2\text{CO}_3 + \text{CaCO}_3 + 4\text{C} = \text{Na}_2\text{Ca}_2\text{C}_4 +$   
 $\text{CaS} + 4\text{CO}$ ;  $\text{Al}_2\text{O}_5 + \text{Na}_2\text{CO}_3 = 2\text{AlNaO}_2 + \text{CO}_2$ ;  $\text{Na}_2\text{CO}_3 +$   
 $\text{Na}_2\text{O}_2 = 2\text{Na}_2\text{O} + \text{CO}_2$ . A yield of 60% Al<sub>2</sub>O<sub>3</sub>  
from a mass. of 0.10% Na<sub>2</sub>O and free from Fe was obtained  
by 10.24 and 11.5 (6.63%), 31.7 Na<sub>2</sub>O (0.9% excess)  
10 g.  $\text{CaCO}_3$  (0.05 excess) and 11 g. C (10% excess)  
else. heating at 1400-1500° for 8 hrs. in a graphite or graphitized  
crucible, heating out the porous melt at 1100° for  
15-20 min. with  $\text{Al(OH)}_3$  with  $\text{C}_6\text{H}_6$  at 100-110° and  
then.

## APPENDIX METALLURGICAL LITERATURE CLASSIFICATION

Reduction of sodium sulfide with hydrogen, methane and water gas. V. P. Il'inskii, O. Ya. Tarasov and A. A. Ginzburg. *Treat. Sulfur Chem. Applied Chem.* (U. S. S. R.) 1962, No. 10, 80-8; cf. Borchakov, C. A. 51, 803; U.S. pat. 3,087,407; Ger. pat. 404,511.---Lab. expts. performed on optimum results of 80% Na<sub>2</sub>S by heating Na<sub>2</sub>S<sub>2</sub> in the presence of 8.6% C catalyst at 200° for 1 hr. with H<sub>2</sub> and water gas and 67% Na<sub>2</sub>S with CH<sub>4</sub>. Al<sub>2</sub>O<sub>3</sub> does not catalyze the reaction. Chas. Black

## ADM-514 METALLURGICAL LITERATURE CLASSIFICATION

1960-1961 1962-1963 1964-1965 1966-1967 1968-1969

Production of carbon dioxide by thermometric method  
V. E. Kabanov, O. Ya. Terskaya and O. Ya. Kabanova  
Trans. Soviet Acad. of Sci., Div. of Phys. Chem. (U. S. S. R.) 1962,  
No. 50, pp. 101-104; *J. Russ. Phys. Chem. C. A.* 19, 3422.—Anhydride  
No. 100,  $\text{Mg}_2\text{O}_3 \cdot \text{MgO}$ , was thermometrically oxidized with charcoal (10  
g. p.  $\text{MgO}$ ) and was thermometrically reduced with charcoal (10  
g. p.  $\text{MgO}$ ) (9 parts of C to 1 part of  $\text{MgO}_2$ ). The anhydride, at  
100-105°, was converted into small bright, sharp-edged, yellow crystals  
of the anhydride, which became larger and brighter, then dried at  
100-105° and heated in a porcelain crucible in an oven at  
100-105° for 2-4 hr. Oxygenous residues were  
removed at 100-105° min. with  $\text{MgO}$  (100 g.  $\text{MgO}$ ) (1-2  
 $\text{MgO}_2$  mass, free 8.0-14,  $\text{MgO}_2$  2.1) and heat, reduced  
100-105°. A smooth oxidation between the conversion of  
unreduced  $\text{MgO}_2$  and  $\text{MgO}_2$  suggests that the initial  
residue from the decomposition of  $\text{MgO}_2$  with  $\text{MgO}$ , which  
in the process of oxidation of the main  $\text{MgO}$  conversion with  
the free 8, and is oxidized to  $\text{MgO}_2$ . By increasing  
the proportion of C, the loss of 8 is increased and decreases  
the proportion of by-products is increased and the yield of  
 $\text{MgO}$  is unchanged. By increasing the temp., to 100°, the  
reaction is complicated, the by-products are depressed and  
the losses increased. An addition of  $\text{K}_2\text{O}$  to the main  
residue is a more complete reduction of  $\text{MgO}_2$  to  $\text{MgO}$ ,  
but does not increase the decomposition of 8, *especially*.  
Chairman

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CH

Production of bromides by the action of bromine on bases in the presence of formate. V. P. Il'inski, A. I. Chertuk and S. L. Rakhmilevich. KHN 1957, 41(1), 29-36. A brief review of methods for the production of bromides. Two methods developed in the U. R. R. R. are described: Il'inski's method. Br<sub>2</sub> is passed into the basic soln., 3Br<sub>2</sub> + 3Na<sub>2</sub>CO<sub>3</sub> + nH<sub>2</sub>O = 3NaBr + NaBrO<sub>3</sub> + 3CO<sub>2</sub> + nH<sub>2</sub>O. The bromate is then wpd. from the soln. by fractional cryst. The soln. after cryst. cryst. bromide and some of the bromate remaining in soln. is converted to bromide by means of iron filings: NaBrO<sub>3</sub> + 2Fe + 3H<sub>2</sub>O = 3Fe(OH)<sub>2</sub> + NaBr. The soln. is filtered free from Fe(OH)<sub>2</sub> and the filtrate evapd. NaBr is centrifuged and dried. The cryst. bromate is recrystallized from water.

6

Il'inski and the soln. added to the bromide soln. and treated with acid: 3NaBr + NaBrO<sub>3</sub> + 3H<sub>2</sub>SO<sub>4</sub> = 3Br<sub>2</sub> + 3Na<sub>2</sub>SO<sub>4</sub> + 3H<sub>2</sub>O. The liberated Br<sub>2</sub> is returned to the process. The second method developed by Chertuk and Rakhmilevich consists in the action of Br<sub>2</sub> on a soln. containing an equim. quantity of alkali and bromate. The reaction takes place in 2 steps: (1) 2NaOH + 2Br<sub>2</sub> + NaBrO<sub>3</sub> + 3H<sub>2</sub>O = 3Br<sub>2</sub> + 2Na<sub>2</sub>SO<sub>4</sub> + NaBrO<sub>4</sub> + 3H<sub>2</sub>O = 7NaBr + 3H<sub>2</sub>O + BrO<sub>4</sub><sup>-</sup>. The other important conditions are: The reaction does not require heating and is conducted at 0°C; it is completed very well at all stages of the reaction and takes about 1.3 hrs.; with a slight excess of formate there are no steps of bromate formed.

REF. 514 METALLURGICAL LITERATURE CLASSIFICATION

Ca

18

New methods for working up sodium sulfite into soda and caustic. V. P. Umnov, N. V. Morin, I. N. Osmirnov, A. R. Kruglov and A. A. Furet. *J. Russ. Ind. (Moscow)* 1944, No. 10, 27-33.—When a Na<sub>2</sub>SO<sub>3</sub> melt, treated at 105° with CO<sub>2</sub>, a max. amt. of Na<sub>2</sub>CO<sub>3</sub> is formed when 70% of the Na<sub>2</sub>SO<sub>3</sub> is decomposed. The impurities are then filtered off, and more CO<sub>2</sub> is passed in to ppt. very pure NaHCO<sub>3</sub>. NaOH is also formed, and H<sub>2</sub>S in the outgoing gases is removed 0.5 hr. after heating the caustic. of CO<sub>2</sub>. Ppt. of NaHCO<sub>3</sub> is washed with added and the melt is boiled for 1.5 hrs. The remaining NaOH is decomposed by this treatment. The CaO content should not exceed more than 2% or calcination of Na<sub>2</sub>CO<sub>3</sub> to Na<sub>2</sub>SO<sub>3</sub> will occur. When the H<sub>2</sub>S content is in the following case is 30% the concn. of CO<sub>2</sub> passed in should be 50%. Larger amounts of CO<sub>2</sub> are undesirable. The optimum concn. of NaOH before the 2nd addition of CO<sub>2</sub> is 184 g. per l. The mother liquors from the process are used for dissolving fresh Na<sub>2</sub>SO<sub>3</sub>. H. M. Lakey

## ASB-36A METALLURGICAL LITERATURE CLASSIFICATION

1810003 MAY ONE ONE

SACRIBED 10

1810003 MAY ONE ONE

SACRIBED 10

Physical-chemical conditions of the production of potassium permanganate. V. P. Il'inski, A. V. Kostylev and N. K. Nefedova. Trudy. Inst. Red. Apat. Chern. U. S. S. R. 19, 8-12 (1954). Numerous right conditions by the electrolytic method show that such production takes place when acid. solns. of  $(\text{NH}_4)_2\text{MnO}_4$  are used. When the soln. becomes alk. slowly (acid.  $\text{KHSO}_4$  is added) tetrahedral. This occurs while the  $(\text{NH}_4)_2\text{MnO}_4$  is added and the resulting  $\text{MnO}_4^-$  precipitates out. The optimum temp. of reaction is 20° and optimum c. of soln. 8 amp./sq. cm. at anode and about 1 at cathode. V. D. Karpovskii

**Sodium sulfide or carbonate.** V. P. Ulinikh and O. Ya. Kalnud. Russ. 42,001, May 31, 1925. (1) melt of Na<sub>2</sub>S obtained in the usual manner from Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, or the cored. soln. of this melt is mixed with mirabilite for the purpose of utilizing the heat of the melt and to decompose the mirabilite. The residue of Na<sub>2</sub>SO<sub>4</sub> is placed into the furnace to be reduced to Na<sub>2</sub>S, and the soln. is concentrated or evaporated.

ASA-SEA METALLURGICAL LITERATURE CLASSIFICATION

Sodium carbonate. V. P. Il'inski and O. Yu. Kolesn.  
Russ. 43,574, Aug. 31, 1938. In the Leblanc soda process  
 $Na_2CO_3$  is melted and by the way NaCl is retained.

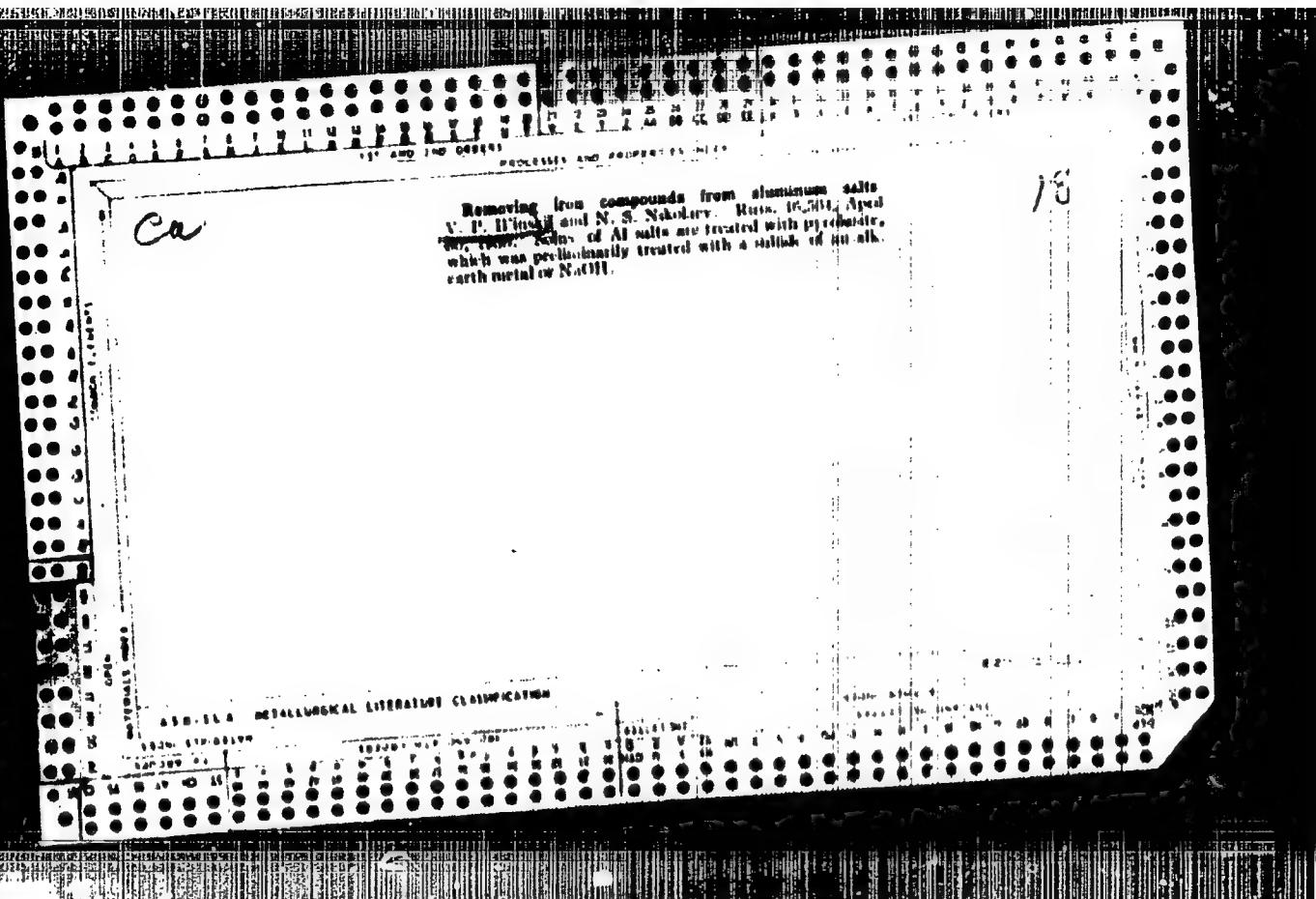
AMERICA METALLURGICAL LITERATURE CLASSIFICATION

18

Sodium hydrosulfide. V. P. Il'inski, A. B. Vinogradov  
and A. S. Isidorov. Russ. Pat. 10,417, April 30, 1980. Bleaching  
 $\text{Na}_2\text{SO}_3 \cdot \text{H}_2\text{O}$  is treated with  $\text{H}_2\text{S}$ .

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ASS-11A METALLURGICAL LITERATURE CLASSIFICATION  
1200 STUDY AREA



en

18  
Anhydrous magnesium chloride. V. P. Il'inskii, V. F. Nekrasov, A. I. Chernik and R. V. Tsvetkov. Izv. Akad. Nauk SSSR, No. 11, 1958. Chlorides of alk. earth metals are added to fused anhyd.  $MgCl_2$  and the result. alk. earth sulfate is separated, by filtration, by settling or by crystallization of  $MgCl_2$ .

AB-11A METALLURGICAL LITERATURE CLASSIFICATION

13  
ca

Thermal insulator. V. P. H'inski. Russ. 47,218, May  
31, 1950. A suspension of asbestos fibers in a MgO  
salt, is heated to form a basic Mg salt and the solid phase is  
separ. and dried.

ASA-15A METALLURGICAL LITERATURE CLASSIFICATION

1800000-1800000

1800000-1800000

READY TO

*ca*

The hydrochemistry of the Karabogaz Bay. V. p. Illinois. D. S. Klybanov and Ya. B. Blyumberg. *Trudy Sel'vost. Lab. (Vsesoyuz. Inst. Galergin)* No. 8, 9-47 (1960).—In the mud of the bay appreciable quantities of  $MgCO_3$  were found. A comparison of the salinity of the waters in the bay in 1967 and 1962 shows a 1.4-fold increase. The increasing of the Mg content of the saline

waters is accounted for by the increased  $MgCO_3$  content of the "white mud." Boring made in 1964-6 proved the presence of mirabilite. Numerous tables on the composition of the saline waters and of the mud at various times during the year are given.  
I. S. Ioffe

410-518 METALLURGICAL LITERATURE CLASSIFICATION

Investigation of the sulfide minerals by the Soviet Academy of Applied Chemistry. V. P. Il'inskii, G. G. Gaidar, *Izv. Akademii Nauk SSSR, Ser. Khim.*, 1939, No. 2, 101-103 (1939); *Khim. Referat. Zhur.*, 1940, No. 2, 101-103. The crystal of mirabilite and thenardite was investigated. Mirabilite can be obtained in amt. equal to 20% of the weight of the brine by natural cooling of the brine from 35° to 5°. A basic method comprises evap. mirabilite by solar evapn., pumping the evapd. brine into a reservoir, and natural cooling to 0°, whereupon mirabilite seps. on the bottom in a layer over 1 m. thick. Then the mother liquor is sent to a reservoir for summer evapn., and seps. of Na<sub>2</sub>SO<sub>4</sub>. Mirabilite can be dehydrated by solar evapn., by heating, by Na<sub>2</sub>S solns., by alk., by NH<sub>3</sub>, by vacuum evapn., and by salting out. Temp. conditions for obtaining NaCl were detd. A new type of rotary furnace for producing Na<sub>2</sub>SO<sub>4</sub> and an electrothermal method for producing NaCl have been developed. The reactions BaS + Na<sub>2</sub>SO<sub>4</sub> + 2H<sub>2</sub>O → BaSO<sub>4</sub> + Na<sub>2</sub>S + H<sub>2</sub>O, and BaSO<sub>4</sub> → BaS were investigated. A new method for producing solid Na<sub>2</sub>SH from Na<sub>2</sub>S, and a double carbonation of NaCl to produce Na<sub>2</sub>CO<sub>3</sub> and S were developed. Conditions for obtaining Na<sub>2</sub>SO<sub>4</sub> sulfate in the rotary furnace were detd. Lab. electrolysis of Na<sub>2</sub>SO<sub>4</sub> to produce NaOH and H<sub>2</sub>SO<sub>4</sub>, and fusion of bauxite with Na<sub>2</sub>CO<sub>3</sub> and C to produce Al<sub>2</sub>O<sub>3</sub>, Na<sub>2</sub>CO<sub>3</sub>, and S were investigated. Com. plants built according to lab. investigations proved very efficient. W. R. Henn

## ADM-15A METALLURGICAL LITERATURE CLASSIFICATION

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12000 STEREOVIEW

S03100 MAT. DIV. 604

CLASSIFIED

CA

Sea water as an industrial source of bromine. V. B. Pinskiy, Trans. Leningrad M. I. Kalinin Polytechn. Inst. 1952, No. 1, 110-21. A new method of extracting salt solutions of low content, including sea water. When carbonates are neutralized with acid, then Br is not fixed by Cl and distilled by means of an air current. The investigation deals with the changing Br content during the operation and with problems of efficiency. Owing to the presence of carbonates, considerable amounts of Cl are rendered useless according to:  $2\text{OH}^- + \text{Cl}^- \rightarrow 2\text{ClO}^- + \text{H}_2\text{O}$ . Using  $\text{H}_2\text{SO}_4$  for pretreatment results in saving Cl. During evaporation, an increasing amt. of Ca and Mg carbonates is paid. The ratio  $\text{Br}^- : \text{H}_2\text{O}_2$  is changing according to from 0.08 at 2% Br. (untreated sea water) to 0.3-0.9 at 30% Br. Consequently the demand for  $\text{H}_2\text{SO}_4$  is reduced 10:1. Evaporation of sea water by radiation is limited to 3% Br per cu. m. because the content of  $\text{MgCl}_2$  is increasing to the point where vapor pressure declines. The content is then 20-30% Br, most of the  $\text{NaCl}$  has already precipitated and the crystall. of  $\text{MgCl}_2 \cdot \text{KCl} \cdot 6\text{H}_2\text{O}$  begins. A higher content is economical only if production of  $\text{MgCl}_2$  is wanted. Data regarding evaporation in different sources are inconsistent. By applying an air current it is possible to produce Br from water having an even lower Br content than sea water. Ed P. Hilpert

The dihydrate of sodium chloride. V. P. Almash, V. P. Korolev, and B. I. Almashov. *Zhur. Khim. i Khim. Prilozhenii* 25, 500-77 (1952) (Eng. translation); *Zhur. Prilozh. Khim.* 25, 307-15 (1952).—The properties of the dihydrate of NaCl formed in nature are discussed, and the production of pure NaCl by means of the dihydrate is investigated. In the temp. interval 0.15 to -21.2°, the stable phase in the system NaCl-H<sub>2</sub>O is the dihydrate. The stability of the dihydrate is detd. by the relative humidity of the surrounding air. In natural conditions in chloride lakes having a small content of SO<sub>4</sub> ions in the brine (value of Jaeger's index for SO<sub>4</sub> is 0.8-3.0), the total salt content frequently exceeds 23.0-24.5%. Consequently with a lowering of temp. in winter it is possible for the dihydrate to crystallize from the brine. Pure NaCl is obtained by means of the dihydrate by cooling a satd. soln. to give max. crystals of the dihydrate, which is then septd. from the mother liquor. At a temp. above 0.15°, the septd. dihydrate breaks up with formation of anhyd. NaCl and a satd. soln. of it. Solns. contg. less than 23.25% NaCl do not sep. the dihydrate on cooling. Calcs. show that by cooling to -21.2°, one metric ton of soln. satd. with NaCl at 25° will sep. 82.0 kg. of the dihydrate, contg. 51.3 kg. NaCl. Also, the quantity and yield of NaCl which is formed on melting of the dihydrate is practically const. within the temp. limits of 0.15-25°. One metric ton of the dihydrate, melting at 25°, yields 481.5 kg. of anhyd. NaCl for a 77.82% yield.

Herbert Liebenland



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SOV/81-59-16-57618

Translation from: Referativnyy zhurnal. Khimiya, 1959, Nr 16, p 281 (USSR)

AUTHORS: Il'inskiy, V.P., Boytsova, V.F.TITLE: The Sorption of Bromine From a Bromine-Air Mixture by Sulfur Dioxide and the Reprocessing of the Obtained Mixture of Acids to  $\text{Br}_2$  or HBr

PERIODICAL: Sb. tr. Gos. in-ta prikl. khimii, 1958, Nr 41, pp 129-152

ABSTRACT: The reaction of interaction of bromine with  $\text{SO}_2$  in the gaseous and liquid phases is studied and the necessary technological indices of the process are obtained. Methods for processing a mixture of HBr, HCl and  $\text{H}_2\text{SO}_4$  acids to liquid bromine or hydrobromic and sulfuric acids have been developed. It has been shown that the reaction of interaction of  $\text{Br}_2$  from a bromine-air mixture with  $\text{SO}_2$  at room temperature (14 - 20°) proceeds quickly and completely at the stoichiometric ratio of  $\text{Br}_2$  and  $\text{SO}_2$ . The sorption of  $\text{Br}_2$  from a bromine-air mixture by an aqueous solution of the mixture of the acids HBr +  $\text{H}_2\text{SO}_4$  runs to completion, but the consumption of  $\text{SO}_2$  is 30 - 35% higher than in the reaction in the gaseous phase. The reaction products are absorbed to 98 - 99% from the gaseous phase by sprinkling the absorption column with a mixture of acids. The concentration of the HBr acid in the solution can be increased to 20% by means of the re-

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SOW/81-59-16-57618

The Sorption of Bromine From a Bromine-Air Mixture by Sulfur Dioxide and the Reprocessing of the Obtained Mixture of Acids to  $\text{Br}_2$  or HBr

circulation of the sorbent in the absorption tower. The vapor pressure of  $\text{HBr} + \text{H}_2\text{SO}_4$  over a mixture of  $\text{HBr} + \text{H}_2\text{SO}_4$  acids for the temperatures 0.25 and 50°C is raised with an increase in the HBr and  $\text{H}_2\text{SO}_4$  concentrations. The absorption coefficient of HBr and  $\text{H}_2\text{SO}_4$ , in the absorption by a mixture of acids, depends principally on the concentration of the sorbent, the temperature and the gas speed and does not depend on the concentration of HBr or  $\text{Br}_2$  in the gaseous phase. The value of the absorption coefficient for the absorption of HBr at a temperature of 14 - 20°C varies from 37 to 40 m/hr and for HCl from 12 to 17 m hr. The oxidation of the acid mixture ( $\text{HBr} + \text{HCl} + \text{H}_2\text{SO}_4$ ) by chlorine to bromine proceeds practically to completion at the introduction of 1.05 - 1.1 chlorine equ. per 1 HBr equ. The distillation of bromine from the solution by steam proceeds to completion at a steam consumption of ~5 kg per 1 kg of bromine; bromine obtained in this way corresponds to the type "oh" (pure). The separation of the acid mixture ( $\text{HBr} + \text{HCl} + \text{H}_2\text{SO}_4$ ) by means of distillation makes it possible to obtain HBr of 40% and  $\text{H}_2\text{SO}_4$  of 70 - 80%. The yield of HBr in the form of a 40% solution is 90 - 95%.

V. Borisova.

Card 2/2

SOW/81-59-16-57619

Translation from: Referativnyy zhurnal. Khimiya, 1959, Nr 16, p 281 (USSR)

AUTHORS: Il'inskiy, V.P., Rusinova, K.D., Drozdova, Ye.O.

TITLE: The Extraction of Bromine by the Method of Air Desorption From High-Thermal Waters

PERIODICAL: Sb. tr. Gos. in-ta prikl. khimii, 1958, Nr 41, pp 153 - 160

ABSTRACT: The oxidation of the  $\text{Br}^-$ -ion in drilling water by chlorine water and gaseous chlorine at an increased temperature ( $70^\circ\text{C}$ ) has been studied. The pressure of  $\text{Br}_2$ -vapor over Cheleken' drilling water at  $65^\circ\text{C}$ , the coefficient of bromine distribution between the gaseous and liquid phases at 25, 40 and  $65^\circ\text{C}$ , and the coefficient of bromine desorption have been determined.

N. Shiryayeva.

Card 1/1

SV/31-59-16-57620

Translation from: Referativnyy zhurnal. Khimiya, 1959, Nr 16, p 281 (USSR)

AUTHORS: Il'inskiy, V.P., Boytsova, V.F., Drozdova, Ye.G., Kuz'mina, N.P., Ruzinova, K.D.

TITLE: The Preparation of Dry Hydrogen Bromide

PERIODICAL: Sb. tr. Gos. in-ta prikl. khimii, 1958, Nr 41, pp 161-170

ABSTRACT: Dry HBr is synthesized from bromine and H<sub>2</sub> in the presence of the "BAU" coal at 600°C; the yield is 91 - 96%. A technological method of purifying and drying HBr has been developed ensuring the preparation of a product containing ~0.04% moisture and H<sub>2</sub>S traces.

N. Shiryayeva.

Card 1/1

SOV/81-59-16-57609

Translation from: Referativnyy zhurnal. Khimiya, 1959, Nr 16, p 280 (USSR)

AUTHORS: Il'inskiy, V.P., Popova, A.V.

TITLE: The Interaction of Chlorides of Alkali Metals With Hydrobromic Acid

PERIODICAL: Sb. tr. Gos. in-ta prikl. khimii, 1958, Nr 41, pp 183-192

ABSTRACT: A new more economic method for obtaining KBr and NaBr has been developed which is based on the interaction of the chlorides of alkali metals and the semi-finished product of bromine production, i.e. solutions of the mixture of the acids HBr + H<sub>2</sub>SO<sub>4</sub> + H<sub>2</sub>O.

N. Shiryayeva.

Card 1/1

SOV/81-59-16-57625

Translation from: Referativnyy zhurnal. Khimiya, 1959, Nr 16, p 282 (USSR)

AUTHORS: Il'inskiy, V.P., Seferovich, Ya.Ye., Uverskaya, A.T., Volnyanskaya, E.M.  
Vysnenkova, O.I.

TITLE: The Preparation of Crystalline Ferrous Bromide by the Sorption of Bromine  
by a Ferrous Bromide Solution

PERIODICAL: Sb. tr. Gos. in-ta prikl. khimii, 1958, Nr 41, pp 193-209

ABSTRACT: Based on the data of the solubility in the system  $FeBr_2 - H_2O$  and thermo-  
chemical calculations on the system  $FeBr_2 - Br_2$  (gas) and  $FeBr_3 - Fe$ ,  
the possibility of obtaining  $FeBr_2$  without smoothing has been proved and  
a method of production has been proposed.

N. Shiriyayeva.

Card 1/1

IL'INSKIY, V.P.; KOKOVKINA, L.I.

Production of potassium and sodium iodides from chlorides and  
hydriodic acid. Med. prom. 13 no.2:20-25 P 159. (MIRA 12:3)

1. Leningradskiy khimiko-farmaceuticheskiy institut.  
(ALKALI METAL IODIDES)  
(HYDRIODIC ACID)

Dissertation: "Vibration Insulation of Industrial Assemblies at the Onset of Dynamic Reactions." Cand Tech Sci, Moscow Inst of Chemical Machine Building, 21 May 54, Vechernaya Moskva, Moscow, 12 May 54.

SO: SUM 284, 26 Nov 1954

IL' INSKIY, V.S.

Protection of laboratory instruments and equipment from vibration.  
Zav.lab.21 no.7:867-868 '55. (MIRA 8:10)

1. Nauchno-issledovatel'skiy institut eksperimental'noy khirurgicheskoy apparatury i instrumentov  
(Measuring instruments) (Vibration)

IL'INSKIY, V.S. kandidat tehnicheskikh nauk.

Vibration isolation in equipment of the chemical industry. Khim.  
prom. no.1:35-41 Ja-F '57. (MLRA 10:4)  
(Vibration) (Chemical engineering--Equipment and supplies)

IL'INSKIY, V.S., kandidat tekhnicheskikh nauk.

Vibration absorbers for machine tool and equipment.  
Mashinostroitel' no.2:31-32 F '57.  
(Machinery--Vibration)

(MLRA 10:5)

IL'INSKIY, V.S.

Vibration dampers used for metal-cutting machine tools. Stan. 1  
instr. 28 no. 5:19-20 Ny '57.  
(Machine tools--Attachments)

117-58-6-16/56

AUTHOR: Il'inskiy, V.S., Candidate of Technical Sciences

TITLE: Devices for Mechanized Cutting, Welding, Soldering, and Metallization (Prisposobleniya dlya mekhanizirovannoy rezki, svarki, payki i metallizatsii)

PERIODICAL: Mashinostroitel', 1958, Nr 6, pp 27-28 (USSR)

ABSTRACT: A device which permits the cutting, welding, soldering, and metallization of cylindrical surfaces, pipes, etc. (Figure 1) is described. The cylindrical surface is held by a roller chain. The spring and the rubber ring of the roller ensure an equal and smooth movement of the device on the cylindrical surface. Another device (Figure 2) is used for form-cutting at the junction of pipes with equal or different diameters, and even of pipes with various details, such as cones, balls, etc. The pipes may be connected at any angle by means of this device. At the same time, a part of the detail may be cut off and an opening made in the other. The cutting speed is regulated by changing the rpm of the drive. The device is of simple construction, light and exact. It was developed by the author. There are 2 figures.

AVAILABLE: Library of Congress  
Card 1/1 1. Machines-Design 2. Cutting tools 3. Welding 4. Soldering

SOV/122-58-8-26/29

AUTHOR: Ilinskiy, V.S., Candidate of Technical Sciences  
TITLE: A Mechanism for Impulsive Feeding of Wire in Apparatus  
for Welding, Cutting, Depositing, Metallising and Soldering  
(Mekhanizm impul'snoy podachi provoloki v apparatakh  
dlya svarki, rezki, naplavki, metallizatsii i payki)

PERIODICAL: Vestnik mashinostroyeniya, 1958, Nr 8, pp 84-85 (USSR)

ABSTRACT: Continuous feeding of wire is said to require bulky mechanism. The impulse principle is claimed to accomplish the feeding with less power and so reduces the size of the mechanism, without impairing the uniformity required in practice owing to the high frequency of the impulses. A rotating, eccentric reciprocates a hollow round slider, surrounding the wire. 3 balls inside the slider wedge the wire by pressing against a conical part of the slider bore when the slider moves forward and release the wire when the slider moves back. During the rearward stroke, the wire remains stationary because it is lightly clamped inside the exit nozzle. The wedging balls can be released by a push button. The rate of wire feed is varied by changing the speed of the eccentric or the amount of eccentricity. Experimental units were made in the shops

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SOV/122-58-8-26/29

A Mechanism for Impulsive Feeding of Wire in Apparatus for Welding, Cutting, Depositing, Metallising and Soldering

of the Vsesoyuznyy nauchno-issledovatel'skiy institut avtogennoy obrabotki metallov (All-Union Scientific Research Institute of Metal Welding) with satisfactory results. There is 1 figure.

1. Welding--Equipment 2. Soldering--Equipment 3. Wire--Handling

Card 2/2 4. Feed mechanisms--Performance

SOV/135-59-1-7/18

AUTHOR: Il'inskiy, V.S., Candidate of Technical Sciences

TITLE: The Optimum Technology of Oxygen Steel Cutting  
(Optimal'nyye rezhimy kislorodnoy rezki stali)

PERIODICAL: Svarochnoye proizvodstvo, 1959, Nr 1, pp 22-24  
(USSR)

ABSTRACT: A series of experiments were carried out to find the optimum technology in oxygen cutting by determining the interdependence of prevalent parameters, such as oxygen consumption, pressure and density and metal thickness. Formulae were developed to compute the cutting rate, and a comparison of results obtained by calculation and experiments showed a satisfactory agreement. The

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SOV/135-59-1-7/18

The Optimum Technology of Oxygen Steel Cutting

optimum parameters for cutting low-carbon steel are given. There are 6 graphs, 1 diagram and 2 tables.

Card 2/2

CLASS I BOOK EXPLOITATION SOV/4377

Il'inskiy, V. S.

Voprosy izolyatsii vibratsii i udarov (Problems in the Isolation of Vibration and Shocks). Moscow, Izd-vo "Sovetskoye radio," 1960. 158 p. No. of copies printed not given.

Ed.: Yu. I. Sukhanov; Tech. Ed.: A. A. Sveshnikov.

PURPOSE: This book is intended for engineers designing radio equipment. It may also be used as a manual for designing vibration-insulating systems.

COVERAGE: The book discusses theoretical and practical problems in protecting various kinds of equipment against vibration and shocks. It deals with a number of currently-used protecting devices and gives such practical details as the data needed to select protective elements for a given equipment. No personalities are mentioned. There are 29 references, 6 Soviet, 21 English and 2 German.

Card 1/4

ZHMUR, A.S.; IL'INSKIY, V.S.; NENYUKOV, V.P.

Single action accelerometers. Izm.tekh. no.12:12-16  
D '62. (MIRA 15:12)  
(Accelerometers)

IL'INSKIY, V. V.

"Investigation of the Effect of Water-Dissolved Air on Cavitation of Hydroturbines." Cand Tech Sci, Khar'kov Polytechnic Inst imeni V. I. Lenin, Min Higher Education USSR, Khar'kov, 1955. (KL, No 15, Apr 55)

SO: Sum. No. 704, 2 Nov 55 - Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (16).

VASIL'YEV, A.; ZAPASNOY, A.; IL'INSKIY, Ye.; PAKUSHIN, V.; SHCHEVCHUK, S.

Business accounting for highway-operation section. Avt.dor.17  
no.1:6 J1-Ag'54. (MIRA 8:10)  
(Roads--Estimates and costs)

IL'INSKIY, YE. M.

Mar 1948

USSR/Mines

Mining Machinery

Mining Methods

"Mining with Open Hoppers in the Temir-Tau Mine and the Bol'shaya Gora Open-Cut Mine,"  
G. A. Gavlich, Ye. M. Il'inskiy, Mining Engineers, 2 $\frac{1}{2}$  pp

"Gornyy Zhur" No 3

Describes in detail the accumulative extraction method of mining used in the Temir-Tau magnetite mine and the Bol'shaya Gora open-cut dolomite mine, and explain the open-hopper systems with aid of diagrams.

PA 51T68

IL'INSKIY, YE. M.

REF ID: A6124

USSR/Mining Methods  
Coal

Sep 48

"Hydraulic Stripping at the Krivoy Rog Coal Field,"  
Ye. M. Il'inskiy, 4 pp

"Mek Trud i Tyazh Rabot" No 9

This operation was first tried at mine No 5 of subject  
coal fields. Briefly describes fundamentals of the  
operation and results. Tables show relative pro-  
duction for 7-month period in 1946. Suggested that  
comprehensive study of actual performance of subject  
method will show even better production results.

28/4/97107

IL'INSKII, Ye.V., kandidat veterinarnykh nauk.

Tissue therapy in surgery. Veterinariia 32 no.9:58-60 8 '55.  
(MINA 8:12)

1.Khar'kovskiy veterinarnyy institut.  
(TISSUE EXTRACTS) (VETERINARY SURGERY)

USSR / General Problems of Pathology. Transplantation U-2  
of Tissues and Tissue Therapy.

Abs Jour: Ref Zhur-Biol., No 15, 1958, 70737.

Author : Ilinckiy, V.  
Inst : Novocherkassk Zootechnical-Veterinary Institute.  
Title : Comparative Evaluation of Certain Methods in  
Tissue Therapy.

Orig Pub: Tr. Novocherkasskogo zootekhn-vet. in-ta, 1957,  
vyp 10, 371-374.

Abstract: Various methods of tissue therapy were used in the treatment of the following: 54 horses with purulent necrosis in the withers region; 73 animals (horses, large horned cattle, dogs and cats) with eye diseases; 81 animals with ulcers; 40 cows with chronic purulent-catarrhal endometritis. The treatment included: 1. implantation of tissues preserved according to Filatov or treated with a two percent solu-

Card 1/2

IL'INSKIY, Ye.V.

Automatic stand tests of mechanisms with pneumatic drives. Avt.  
(MIRA 13:8)  
prom. no.6:29 Je '60.

1. Lvovskiy avtobusnyy zavod.  
(Motorbuses--Pneumatic equipment)  
(Electronic instruments)

S/100/61/000/002/003/010  
B113/B203

9.2572

AUTHORS:

Il'inskiy, Yu.A.; Karasev, M.D.

TITLE:

Study of transition processes in a two-circuit parametric transformer with sum and difference output frequency

PERIODICAL: Vestnik Moskovskogo universiteta. Seriya 3, fizika, astronomiya, no. 2, 1961, 12 - 18

TEXT: First, the authors analyze the transition processes. According to Ref. 1 (Karasev, M.D. UFN, LXIX, vyp. 2, 1959), a two-circuit parametric transformer can be described by a system of equations, Eq. (1)  $\ddot{x}_i + \omega_i^2 x_i - 2\delta_i x_i + a_i f(\omega t)(x_1 + x_2) + e_i$ . The frequency of the parametric change  $\omega$  by the relation  $\omega = \omega_2 - \omega_1 + \Delta$  is related to the circuit frequencies. The upper sign holds for a rotating (difference) modulator, the lower one for a non-rotating (sum) modulator;  $\Delta$  is a slight mis-tuning. The external force is supposed to be introduced in one circuit only so that the slowly variable, complex amplitude  $E_2$  of the external

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S/188/61/000/002/003/010  
B113/B203

Study of transition processes ...

force vanishes and the frequency  $\rho = \omega_1 + \lambda_1$ . The damping factor of oscillations after switching off the external signal is obtained from Eq. (2)

$$x_1 = |z_1^{(1)}| e^{\rho_1 t} \cos \left[ (\omega_1 + \frac{\Delta}{2} + i m \lambda_1) t + \arg z_1^{(1)} \right] +$$

$$+ |z_1^{(2)}| e^{\rho_2 t} \cos \left[ (\omega_1 + \frac{\Delta}{2} + i m \lambda_2) t + \arg z_1^{(2)} \right].$$

$$x_2 = |z_2^{(1)}| e^{\rho_1 t} \cos \left[ (\omega_1 \mp \frac{\Delta}{2} + i m \lambda_1) t + \arg z_2^{(1)} \right] +$$

$$+ |z_2^{(2)}| e^{\rho_2 t} \cos \left[ (\omega_1 \mp \frac{\Delta}{2} + i m \lambda_2) t + \arg z_2^{(2)} \right].$$

$$z_1^{(1)} = \frac{E}{2i\omega_1(\lambda_1 - \lambda_2)} \frac{1}{\lambda_1 + i \frac{\Delta}{2} - i \Delta_1}$$

$$z_2^{(1)} = \mp \frac{E}{2i\omega_1(\lambda_1 - \lambda_2)} \frac{\text{Im} z_1}{2\omega_1} \frac{1}{\lambda_1 + i \frac{\Delta}{2} - i \Delta_1}$$

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B113/B203

Study of transition processes ...

and the formation of oscillations after switching on the external force proceeds by the law  $x'_1 = x_1^{(0)}(t) - x_1'(t)$ , the steady solution  $x_1^{(0)}$  having the form of Eq. (3)

$$x_1^{(0)} = |x_1^{(0)}| \cos(\phi t + \arg x_1^{(0)}),$$

$$x_1' = |x_1^{(0)}| \cos((\omega \mp \rho)t \mp \arg x_1^{(0)}),$$

$$x_1^{(0)} = \frac{B}{2j\omega_1} \frac{\omega_1 - j(\Delta - \Delta_1)}{\left(\omega_1 + j\frac{\Delta}{2} - j\Delta_1\right)\left(\omega_1 + j\frac{\Delta}{2} + j\Delta_1\right)}$$

$$= \frac{B}{2j\omega_1} \frac{\omega_1 - j(\Delta - \Delta_1)}{(\omega_1 + j\Delta_1)(\omega_1 - j(\Delta - \Delta_1)) \mp \frac{\omega_1 \Delta_1}{4\omega_1 \omega_2} |f_1|^2}.$$

$$x_1^{(0)} = \pm \frac{B}{2j\omega_1} \frac{j\omega_1 \mp 1}{2\omega_2} \frac{1}{\left(\omega_1 + j\frac{\Delta}{2} - j\Delta_1\right)\left(\omega_1 + j\frac{\Delta}{2} + j\Delta_1\right)}$$

$$= \pm \frac{B}{2j\omega_1} \frac{j\omega_1 \mp 1}{2\omega_2} \frac{1}{(\omega_1 + j\Delta_1)(\omega_1 - j(\Delta - \Delta_1)) \mp \frac{\omega_1 \Delta_1}{4\omega_1 \omega_2} |f_1|^2}.$$

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S/100/61/000/002/003/010  
B113/B203

## Study of transition processes ...

The trajectories of the roots  $\lambda_{1,2}$  characteristic of transition processes and steady behavior having the form of Eq. (A)

$\lambda_{1,2} = -\frac{\delta_1 + \delta_2}{2} \pm \sqrt{\frac{1}{4}(\delta_1 - \delta_2 + j\Delta)^2 + \frac{1}{4\omega_1\omega_2} |f_1|^2}$ , where  $f_1$  is determined by the Fourier expansion of the function  $f(\omega t) = \sum f_k e^{j\omega_k t}$ , are shown in Figs. 1 and 2 in the complex plane with a change in the parameter  $\Delta$  and  $k^2 = \frac{\alpha_1 \alpha_2}{4\omega_1 \omega_2} |f_1|^2$ . The trajectories of  $\lambda_2$  are on the left, those of  $\lambda_1$  on the right of  $-\frac{\delta_1 + \delta_2}{2}$ ; the trajectories of the rotating modulator

are within  $-\delta_2$  and  $-\delta_1$ , those of a non-rotating modulator are without. On the basis of Eq. (2) and Eq. (3) and Figs. 1 and 2, the following has been found: In a rotating modulator, the transition process is the sum of two damped harmonic oscillations (with switched-off external signal), or the sum of two damped and one steady oscillation (with switched-on exter-

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S/188/61/000/002/003/010  
B113/B203

Study of transition processes

nal force). For a non-rotating modulator, it follows from Eq. (3), that the frequency characteristic of the amplifier is represented by the product of the frequency characteristics of individual circuits. Experiments dealt with a low-frequency parametric transformer with circuit frequencies of about 100 and 400 kc/sec. In conformity with Eq. (2), the transition processes had two constant times, and pulsations in the transition processes were observed in the case of mistuning. Further, it was found that the duration of the transition process, in the case of large K, was proportional to K (K being the voltage amplification factor in the first circuit of a regenerative transformer). The duration of the transition process was the time during which the oscillation amplitude attained the  $(1 - e^{-1})$ -fold of its steady value. In a non-rotating modulator, the duration of the transition process dropped monotonically with increasing  $k^2$  in both circuits. The time of formation in the second circuit was slightly longer than in the first one due to a slow increase of oscillations in the second circuit at the beginning of the formation process. There are 6 figures and 1 Soviet-bloc reference.

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8/188/61/000/002/003/010

B113/B203

Study of transition processes

ASSOCIATION: Kafedra teorii kolebanii (Department of the Theory of Oscillations)

SUBMITTED: June 19, 1960

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S/188/61/000/006/001/007  
B108/B138

9,2572

AUTHORS: Il'inskiy, Yu. A., Karasev, M. D.

TITLE: Double-tuned parametric oscillator with external feed

PERIODICAL: Moscow Universitet. Vestnik. Seriya III. Fizika, astronomiya, no. 6, 1961, 3 - 11

TEXT: A double-tuned parametric amplifier operates as an oscillator if the feed amplitude is high enough. A self-excited oscillator whose amplitude is limited by a non-linear resistor in one of its circuits is considered. Synchronization of the oscillator with an external force is also dealt with. In experimental investigations, the authors used a parametric oscillator with frequencies of 100 and 400 kcps in the two circuits, with semiconducting diodes of the types Д1Ч-27 (DGTs-27) and Д809 (D809) serving as nonlinear capacity. It was found that the amplitude is limited either by a nonlinear resistance or by a dependence of the parametric connection between the circuits on amplitude fluctuations. Perturbations that are due to the nonlinear reactance lead to a

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Card 1/2

Double-tuned parametric...

S/188/61/000/006/001/007  
B108/B138

distortion of the amplitude-frequency characteristics. An external force produces the same synchronization and beat phenomena as in a tube generator. In the r.f.-range, calculated and experimental results are entirely consistent. There are 6 figures and 6 Soviet references.

ASSOCIATION: Kafedra teorii kolebaniy (Department of the Theory of Oscillations)

SUBMITTED: January 30, 1961

VB

Card 2/2

24895

9.2572  
S/109/61/006/008/015/018  
D207/D304

AUTHORS: Il'inskiy, Yu.A., and Karasev, M.D.

TITLE: Transients in a two circuit parametric amplifier

PERIODICAL: Radiotekhnika i elektronika, v. 6, no. 8, 1961,  
1397 - 1400

TEXT: The authors give a short report on the theoretical and experimental analysis of transients in a two circuit parametric amplifier. The theoretical analysis is carried out assuming a small signal. The equivalent circuits of two circuit parametric amplifiers are shown in Fig. 1. These are systems with two degrees of freedom and with periodically varying reactive element. The equations for the above systems may be written as

$$\ddot{x}_i + \omega_i^2 x_i = \mu (-2\delta \dot{x}_i + \omega_i / (m) (x_1 + x_2) + e_i) \quad (1)$$

where  $i = 1, 2$  the index related to either the first or the second

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Transients in a two circuit ...

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S/109/61/006/008/015/028  
D207/D304

circuit;  $e_1$  - external driving force;  $e_2 = 0$ ;  $\mu$  - a small parameter;  $\sigma$  - attenuation of the  $i$ -th circuit;  $f$  - a periodic function with period 2;  $\omega$  - the frequency of change of the parameter. It follows from Eq. (1) that the cct attenuation and modulation depth are assumed to be small, so that the Q of the circuits is high and parametric coupling weak; oscillations in the system are very nearly harmonic. The constant coupling may be large. Eq. (1) can be solved by one of the quasi-linear methods, e.g. by averaging. Eq. (1) is first reduced to standard form by assuming  $\omega = \omega_1 + \omega_2$  and putting

$x_i = Z_i e^{j\omega_i t} + Z'_{-i} e^{-j\omega_i t}$ ,  $x_i' = j\omega_i (Z_i e^{j\omega_i t} - Z'_{-i} e^{-j\omega_i t})$ ,  
where  $i = 1, 2$ ;  $Z'_{-i} = Z_i^*$  - complex amplitudes, so that after averaging the shortened form

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S/199/61/006/008/015/018,  
D207/D304

Transients in a two circuit ...

$$\dot{z}_1 = \mu \left\{ - (S_1 + j\Delta_1 + \frac{j\alpha_1 f_0}{2\omega_1}) z_1 - \frac{j\alpha_1 f_1}{2\omega_1} z_{-2} + \frac{E}{2j\omega_1} \right\}, \quad \dot{z}_{-2} = \mu \left\{ - (\delta_2 - j\Delta_2 - \frac{j\alpha_2 f_0}{2\omega_2}) z_{-2} + \frac{j\alpha_2 f_{-1}}{2\omega_2} z_1 \right\} \quad (2)$$

is obtained where  $f_0, f_1, f_{-1}$  - the Fourier coefficients of function  $f(\omega t)$

$$f(\omega t) = \sum_{k=-\infty}^{k=+\infty} f_k e^{j\omega kt}$$

It is assumed further that  $f_0 = 0$  since

$$\Delta_1 = \Delta_1 + \frac{\alpha_1 f_0}{2\omega_1}$$

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Transients in a two circuit ...

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S/109/61/006/008/015/018  
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can be introduced and the detuning due to the constant component of coupling thus taken into account, the solution of linear equations (2) with constant coefficients is easily found for any external force  $E$  and any initial conditions. E.g. the transient state can be determined, i.e. the presence of oscillations in the system; when  $E = \text{constant}$  and initial condition  $Z_1(0) = 0$  the equations have then the shape of matrix

$$\begin{aligned} \dot{Z} &= AZ + B, \\ Z &= \begin{pmatrix} Z_1 \\ Z_2 \end{pmatrix}, \quad B = \mu \frac{E}{2j\omega_1} \begin{pmatrix} 1 \\ 0 \end{pmatrix}. \end{aligned}$$

Its solution, satisfying the initial conditions  $Z(0) \neq 0$  is

$$Z = \int_0^t e^{A(t-s)} B(s) ds = \begin{pmatrix} 1 \\ \mu^2 (\lambda_2 - \lambda_1) \end{pmatrix} \left\{ \frac{1 - e^{\mu \lambda_1 t}}{\lambda_1} [AB - \lambda_1 B] - \frac{1 - e^{\mu \lambda_2 t}}{\lambda_2} [AB - \lambda_2 B] \right\} \quad (5)$$

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Transients in a two circuit ...

S/1C9/61/006/008/015/018  
D207/D304

It follows that in a two circuit parametric amplifier the time of establishment of the steady state does not depend on the relationship between the phase of the suddenly applied signal and the phase of pulsed variation of the parameter; that at optimum tuning ( $\Delta_1 = \Delta_2 = 0$ ) the characteristic roots are real. With  $f_1 \neq 0$  (no varying parameter) these are essentially equal to  $-\delta_1$  and  $-\delta_2$  and diverge symmetrically with respect to the centre of  $(-\delta_2, -\delta_1)$  with increasing  $|f_1|$ . When the smaller of the roots (absolute value) reaches zero the system becomes self oscillating. When this root assumes smaller absolute values then the duration of the transient process increases. Finally with the detuning of the sets ( $\Delta_1$  or  $\Delta_2 \neq 0$ ), then  $\Im \lambda_{1,2} \neq 0$  and the transient contains damped oscillations. The theory was experimentally applied to a balanced parametric amplifier with its first set tuned to 100 Kc/s and its second set to 100 Kc/s.

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9,2580 (230)

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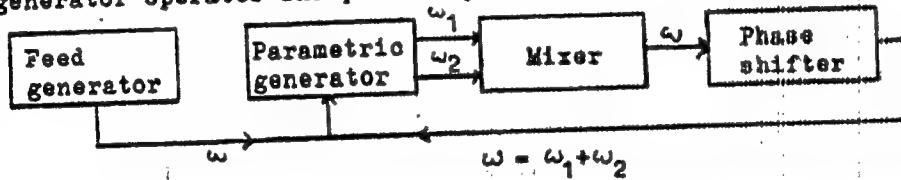
S/18B/62/000/002/008/013  
B154/B102

AUTHOR:

Il'inskiy, Yu. A.

TITLE:

About a two-circuit parametric generator

PERIODICAL: Moscow. Universitet. Vestnik. Seriya III. Fizika,  
astronomiya, no. 2, 1962, 60-67TEXT: For rational ratios  $\omega_1/\omega_2$ , the oscillations (frequencies  $\omega_1$  and  $\omega_2$ ) generated in a two-circuit generator by a frequency  $\omega$  are investigated. The author first demonstrates the existence of a range in which the ratio  $\omega_1/\omega_2$  is rational for  $\omega_1 + \omega_2 = \omega$ . In the arrangement shown the generator operates independently of an external source.

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S/188/62/000/002/008/013

B154/B102

About a two-circuit parametric ...

When the relation between the natural frequencies  $\omega_{10}$  and  $\omega_{20}$  of the circuits is  $\omega_{20} \approx n\omega_{10}$ , the oscillations can be synchronised by a single external voltage, since also the phases are related by  $n$ . The frequency of this synchronizing voltage has to be close to the natural frequency of one of the circuits. Measurements of the amplitude  $E_1$  of the synchronizing voltage made at  $\omega_{20} \approx 3\omega_{10}$  and  $102.3 \text{ kc} \leq \frac{\omega_1}{2\pi} \leq 103.2 \text{ kc}$  show that the frequencies  $\omega_1$  and  $\omega_2$  are synchronized ( $\omega_2 = 3\omega_1$ ). In this case the range of synchronization virtually does not depend on  $E_1$ , because it is determined by the oscillations themselves and the oscillations essentially do not depend on the synchronizing voltage. Influences on the oscillations by the synchronizing voltage and an expansion of the synchronization band with increasing  $E_1$  were only observed in a few cases. The investigations show that a generator with external excitation can be used as an effective frequency splitter. The band width is maximum if the quality factors of the circuits are almost equal. There are 9 figures.

4

Card 2/3

About a two-circuit parametric ...

S/188/62/000/002/008/013  
B154/B102

ASSOCIATION: Kafedra teorii kolebaniy, Moskovskiy universitet  
(Department for the Theory of Oscillations, Moscow  
University)

SUBMITTED: July 3, 1961

Card 3/3

IL'INSKIY, Yu.A.

Effect of field gas processing on the efficiency of gas transportation. Gaz. delo no. 7:30-33 '65. (MIRA 18:9)

1. Khar'kovskoye upravleniye magistral'nykh gasoprovodov.

IL'INSKII, Yu.A.

Clinical aspects and therapy of protosan colitis. Klin. med., Moskva 31  
no.6:63-68 June 1953. (GMML 25:1)

1. Sixth Course Therapeutic Faculty Student. 2. Of the Clinic for In-  
fectious Diseases (Director -- Prof. A. V. Bilibin), Second Moscow Medi-  
cal Institute imeni I. V. Stalin.

## EXCERPTA MEDICA Sec 6 Vol 13/2 Internal Med. July 56

## 3427. INVESTIGATION OF THE DISTURBANCE OF THE CARBOHYDRATE FUNCTION OF THE NEURO-GLANDULAR SYSTEM OF THE LIVER IN EPIDEMIC HEPATITIS (Russian text) - Ilinskij, Yu. A. Dept. of the Pirogov Second Med. Inst. of Infect. Dis., Moscow - SBORN. 'BOLEZN BOTKINA' (Moskva) (Eds: Tareeva E. M. and Shubladze A. K.) 1956 (235-244)

The carbohydrate function of the liver was studied in healthy subjects and in patients with various forms of epidemic hepatitis using Speranaki's method which is based on the application of stimulants of varying strength (40 ml. and 5 ml. 40% i.v. glucose solution). In the healthy subjects the curve obtained corresponded exactly to the strength of the stimulant applied. In the 86 investigated patients with epidemic hepatitis (16 mild, 52 moderately severe and 18 severe) the responses did not correspond to the strength of the stimulant. In some cases at the peak of the illness identical or similar rises of the glycaemic curves were observed as a response to the strong as well as to the weak stimulant; in other cases the rise in response to the weak stimulant was higher than to the strong; a negligible rise of the glycaemic curves in response to the strong or either stimulant was observed in isolated cases. These results are to be regarded as an expression of plastic conditions of the neuro-glandular system of the liver - the compensating, paradoxical and inhibitory stages. Differentiated responses corresponding to the strength of the applied stimulant were observed in mild cases of infectious hepatitis and in the period of remission but with a higher hyperglycaemic coefficient than in the healthy subjects. In cases of recovery with residual symptoms a differentiated response to the i.v. injection of stimulants of varying strength was present for a long time, with high hyperglycaemic coefficients.

Guseva - Moscow (S)

USSR/Virology - Viruses of Man and Animals.  
Viruses of Hepatitis.

E

Abs Jour : Ref Zhur Biol., No 6, 1959, 23890

Author : Il'inskiy, Yu.A.

Inst : Moscow Medical Institute

Title : Some Colloid Reactions of the Blood in Botkin's Disease

Orig Pub : Uch. zap. 2-y Mosk. med. in-t, 1957, 7, 53-57

Abstract : No abstract.

Card 1/1

KRASNOGOLOVETS, V.N.; IL'INSKIY, Yu.A.

Colimycin and mycerin therapy [with summary in English]. Antibiotiki 3 no.6:102-107 N-D '58. (MIRA 12:2)

1. Klinika infektsionnykh bolezney (zav. - chlen-korrespondent AMN SSSR prof. A.P. Bilibin) II Moskovskogo meditsinskogo instituta imeni N.I. Pirogova.

(DYSENTERY, BACILLARY, ther.  
colimycin & mycerin (Rus))

(ANTIBIOTICS, ther. use,  
colimycin & mycerin in dysentery (Rus))

IL'INSKIY, Yu.A. (Moskva)

Pneumonia in ornithosis. Klin.med. no.12:196-59 '61,

(MIRA 15:9)

1. Iz kafedry infektsionnykh bolezney (zav. - deyствител'nyy  
chlen AMN SSSR prof. A.F. Bilibin) II Moskovskogo meditsinskogo  
instituta imeni N.I. Pirogova.  
(PNEUMONIA) (ORNITHOSIS)

IL'INSKIY, Yu.A.; MARKOVA, Ye.A. (Moskva)

Forme fruste of Botkin's disease. Klin.med. 39 no.3:46-50  
Mr '61. (MIRA 14:3)

1. Iz kliniki infektsionnykh bolezney (dir. - deyствител'nyy  
chlen AMN SSSR prof. A.P. Bilibin) II Moskovskogo meditsinskogo  
instituta imeni N.I. Pirogova.  
(HEPATITIS, INFECTIOUS)

BILIBIN, A.F., prof.; IL'INSKIY, Yu.A., kand.med.nauk

Epidemiology and clinical aspects of ornithosis. Klin.med. no.3:  
34-38 '62. (MIRA 15:3)

1. Deystvitel'nyy chlen AMN SSSR (for. Bilibin).  
(ORNITHOSIS)

BILIBIN, A.F., prof., red.; IL'INSKIY, Yu.A., red.

[Clinical problems of infectious pathology] Klinicheskie voprosy infektsionnoi patologii. Pod red. A.F.Bilibina. Moskva, Meditsina, 1965. 275 p.  
(MIRA 18:12)

1. Moscow. Vtoroy meditsinskiy institut. 2. Deyatel'nyy chlen AMN SSSR (for Bilibin).

IL'INSKIY, YU. A.

USSR/Medicine - Therapeutic Sleep, Narcosis

Jul/Aug 51

"Treatment With Prolonged Sleep Brought About by Introducing Intravenously an Alcohol-Containing Solution By the Continuous Drip Method," O. V. Kurbikov, Ye. S. Zorina, Yu. A. Il'inskij

"Nevropatol i Psichiat" Vol XX, No 4, pp 38-40

Describes clinical aspects of treating psychopathic patients by intravenous introduction of alc and technique of introducing a narcotic mix which has the following compnt: sodium chloride 4.0, calcium chloride 1.0, glucose 25.0, distilled alc 60.0-120.0, distilled water up to 500.0

198758

IL'INSKIY, Yu. A.

KERBIKOV, O. V., ZORINA, YE. S., IL'INSKIY, YU. A.

Blood - Analysis and Chemistry

Concerning Prof. Ye. Yu. Karu's remarks "On the determination of alcohol in the blood by the Vidmark method." Zhur. nevr, i psikh, 52, no. 3, March 1952.

Monthly List of Russian Accessions, Library Of Congress, August, 1952. Unclassified

IL'INSKIY, Yu. A.

"Clinicoexperimental Data Concerning the Development of One of the Sleep Therapy Methods (The Treatment of Psychotic Patients by Prolonged Sleep Produced by the Intravenous Drip of an Alcohol-Containing Solution)." Cand. Med. Sci., Moscow State Medical Inst, Moscow, 1953. (RZhBiol, No 7, Doc 54)

Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (12)  
SO: Sum. No. 556, 24 Jun 55

ZHARIKOV, N.M.; IL'INSKIY, Yu.A.; KERBIKOV, O.V.; MATVYETS, I.S.

Data on immunological reactivity in schizophrenia. Zhur.nevr. i psich.  
(MLRA 9:11)  
56 no.8:612-621 '56.

1. Kafedra psichiatrii II Moskovskogo meditsinskogo instituta (zav. -  
prof. O.V.Kerbikov) i laboratoriya tulyaremii (zav. - prof. N.G.  
Olsuf'yev) Instituta epidemiologii i mikrobiologii imeni N.F.Gamalei  
ANN SSSR, Moskva.  
(SCHIZOPHRENIA, immunology,  
(Rus))

*Л. ИНСКИЙ, Ю.А.*

Struggle against refusal of food and emaciation in psychiatric patients. Zhur.nevr. i psikh. Supplement:91-92 '57. (MIRA 11:1)

1. Kafedra psichiatrii (sav. - prof. O.V.Kerbikov) II Moskovskogo meditsinskogo instituta imeni I.V.Stalina.  
(PSYCHOSES) (APPETITE) (ALCOHOL--THERAPEUTIC USE)

IL'INSKIY, Yu.A.

"Schizophrenia" [in English and German] by Derek Richter.  
Reviewed by Iu.A.Il'inskii. Sov.med. 23 no.6:156-157  
Je '59. (NIBA 12:9)

(SCHIZOPHRENIA) (RICHTER, DEREK)

IL'INSKIY, Yu.A.; KULIKOV, L.S.

Immunological reactivity of mental patients following therapy  
with aminazine, insulin, and other methods [with summary in French].  
Zhur.nevr. i psich. 59 no.2:156-159 '59. (MIRA 12:4)

1. Kafedra psichiatrii (zav. - prof. O.V. Kerzhikov) II Moskovsko-  
go meditsinskogo instituta im. N.I. Pirogova.  
(MENTAL DISORDERS, immunol.  
off. of ther. (Rus))

IL'INSKIY, Yu.A.

Allergic reactivity of patients with schizophrenia. Zhur. nevr. i  
psikh. 61 no.4:549-556 '61. (MIRA 1417)

1. Kafedra psichiatrii (sav. - prof. O.V.Kerbikov) i kafedra patologicheskoy fiziologii (sav. - prof. A.D.Ado) II Moskovskogo meditsinskogo instituta imeni N.I.Pirogova. (SCHIZOPHRENIA) (ALLERGY)

IL'INSKY, Yu.A.

Changes in some allergic indices in the treatment of schizophrenia patients. Probl. sud. psich. no.13:74-84 '62. (MIRA 18:9)

IL'INSKIY, Yu.A., kand. med. nauk

Antistaphylococcal immunity in schizophrenia. Vrach. delo no.11:  
79-82 N°63 (MIR 16:12)

1. Kafedra psikiatrii (zav. — deystvitel'nyy chlen AMN SSSR,  
prof. O.V. Kerbikov) 2-go Moskovskogo instituta.

IL'INSKIY, Yu.A.

Immunological reactivity of patients with schizophrenia during reserpine treatment. Vest. AMN SSSR 17 no.1:43-51 '62. (MIRA 15:3)

1. Is kafedry psikiatrii (zav. - chlen-korrespondent AMN SSSR prof. O.V. Kerbikov) II Moskovskogo meditsinskogo instituta imeni N.I. Pirogova.

(RESERPINE) (SCHIZOPHRENIA)  
(IMMUNOLOGY)

5 (2), 5 (4)

AUTHORS:

Shchukarev, S. A., Oranskaya, M. A., Tolmachova, T. A.,  
Il'inskii, Yu. S.

SOV/78-5-1-2/45

TITLE:

Thermal Dissociation of Vanadium Dichloride

PERIODICAL:

Zhurnal neorganicheskoy khimii, 1960, Vol 5, Nr 1, pp 8 - 11  
(USSR)

ABSTRACT:

Publications give different data for the formation enthalpy  $\Delta H$  of  $VCl_2$ . The authors report on their indirect determination of  $\Delta H$  by investigation of the equilibrium of  $VCl_2$  reduction by means of H at  $750^\circ$ ,  $775^\circ$ ,  $800^\circ$ , and  $825^\circ$ . The method is described in references 9,10. The experiments lasted for 100-200 hours. Table 1 shows the values of the dissociation pressure of  $VCl_2$ . Figure 1 shows the linear dependence of  $\lg P_{VCl_2}$  on  $\frac{1}{T}$ . The computed values of the formation enthalpy  $\Delta H$  and of the absolute entropy  $\Delta S$  are shown in table 2. The value found for  $\Delta H$  is in good agreement with that assumed by the U.S.A. Bureau of Standards. Figure 2 and table 3 show the opposite behavior of

Card 1/2

Thermal Dissociation of Vanadium Dichloride

SOV/78-5-1-2/45

the dissociation enthalpy on the one hand and of the sum of the two ionization potentials and the sublimation energy on the other hand in the case of elements with the atomic numbers 21 - 30. There are 2 figures, 3 tables, and 18 references, 9 of which are Soviet.

SUBMITTED: October 27, 1958

Card 2/2

IL'INYKH, A., insh.

Modernizing a "rondel" machine. Prom.koop. 13 no. 5:12  
My '59. (MIRA 12:9)

1. Oblpromsovet, g. Perm.  
(Pearl button industry--Equipment and supplies)

ZININ, V.P.; IL'DYIN, A.P.

Automatic electrically operated ventilation door. Trudy Unipromed  
no.2:174-182 '57.  
(Mine ventilation) (Automatic control)

PETROV, I.P., dotsent; IL'INYKH, A.F., inzh.

Multichannel TU-TS system with polarity-amplitude block selection. Izv.vys.ucheb.zav.; gor.zhur. no.4:61-86 '59.  
(MIRA 13:5)

1. Sverdlovskiy gornyy institut imeni V.V. Vakhrusheva.  
Rekomendovana kafedroy gornoj elektrotehniki.  
(Mine communications)  
(Remote control)  
(Telemetering)

IL'INYKH, A.F., inzh.

Selection of an efficient TU-TS system for remote control of substations. Izv.vys.ucheb.zav.; gor.shur. no.31:33-138 '61.  
(MIRA 15:4)

1. Sverdlovskiy gornyy institut imeni V.V.Vakhrusheva; rekomendovana kafedroy obshchey elektrotekhniki Sverdlovskogo gornogo instituta.  
(Mine communications) (Remote control)

APSHINSKIY, V.M.; BACAUTINOV, G.A.; BESPALOV, M.V.; GASPAROVICH, P.I.;  
GOLOMIDOV, I.N.; GOLUBOV, G.B.; GRIN, L.T.; ZEL'SKIY, S.A.;  
JL'INYKH, A.F.; KOZIN, V.Z.; KRYUKOV, V.P.; KULAKOV, S.N.;  
LUKAS, V.A.; MINEYEV, V.A.; PETROV, Yu.S.; PIKUSHKO, M.G.;  
PROKOF'YEV, Ye.V.; REBETS, B.A.; STARTSEV, N.V.; THOP, A.Ye.,  
prof.; KHRAMOV, V.A.; ABRAMOV, V.I., otv. red.; PROZDROVSKAYA,  
V.L., tekhn. red.; BOLDYREVA, Z.A., tekhn. red.

[Handbook on electric equipment for mines] Sjpravodchik gorno-  
go elektrotekhnika. Pod obshchei red. A.E.Tropa. Moskva,  
Gosgortekhizdat, 1962. 400 p. (MIRA 16:5)  
(Electricity in mining)

RUDNEV, D.F.; IL'INYKH, A.I.

Results of using DDT solution in diesel fuel to combat orchard pests.  
Nauch.trudy Inst. ent. i fit. 6:89-91 '55. (MLRA 9:7)  
(DDT (Insecticide)) (Fruit--Diseases and pests)

IL'INYKH, A.P.

Effect of ACTH and cortisone on the blood serum proteins and  
soluble liver proteins in experimental tuberculosis. Probl.  
tub. 42 no.10:69-74 '64. (MERA 18:11)

1. Biokhimicheskaya laboratoriya (rukoveditel' - dotsent V.A.  
Shcherbatskaya) Sverdlovskogo nauchno-issledovatel'skogo  
instituta tuberkuleza (direktor - prof. I.A. Shakhlein).

IL'INYKH, A.P.

Liver proteins in the development of tuberculosis in rabbits.  
Pat. fiziol. i eksp. terap. 8 no.1:40-43 Ja-F '64.

(MIRA 1812)

1. Biokhimicheskaya laboratoriya "rukovoditel' V.A. Shcherbatakaya)  
Sverdlovskogo nauchno-issledovatel'skogo instituta tuberkuliza (dir.-  
prof. I.A. Shklein) Ministerstva zdravookhraneniya RSFSR.